

# Hedycariol

<b>Inchi:</b>	InChI=1S/C15H26O/c1-12-6-5-7-13(2)9-11-14(10-8-12)15(3,4)16/h6,9,14,16H,5,7-8,10-1
<b>InchiKey:</b>	SDMLCXJKAYFHQM-AAVBEOHCSA-N
<b>Formula:</b>	C15H26O
<b>SMILES:</b>	CC1=CCC(C(C)(C)O)CCC(C)=CCC1
<b>Mol. weight [g/mol]:</b>	222.37

## Physical Properties

Property code	Value	Unit	Source
gf	-41.85	kJ/mol	Joback Method
hf	-391.61	kJ/mol	Joback Method
hfus	16.38	kJ/mol	Joback Method
hvap	67.39	kJ/mol	Joback Method
log10ws	-4.84		Crippen Method
logp	4.230		Crippen Method
mvol	208.620	ml/mol	McGowan Method
pc	2066.12	kPa	Joback Method
rinpol	1671.00		NIST Webbook
tb	676.46	K	Joback Method
tc	889.94	K	Joback Method
tf	341.91	K	Joback Method
vc	0.756	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	599.21	J/molxK	676.46	Joback Method
cpg	619.44	J/molxK	712.04	Joback Method
cpg	638.35	J/molxK	747.62	Joback Method
cpg	655.96	J/molxK	783.20	Joback Method
cpg	672.31	J/molxK	818.78	Joback Method
cpg	687.41	J/molxK	854.36	Joback Method
cpg	701.30	J/molxK	889.94	Joback Method
dvisc	0.0069347	Paxs	341.91	Joback Method
dvisc	0.0011570	Paxs	397.67	Joback Method

dvisc	0.0002998	Paxs	453.43	Joback Method
dvisc	0.0001044	Paxs	509.19	Joback Method
dvisc	0.0000448	Paxs	564.94	Joback Method
dvisc	0.0000224	Paxs	620.70	Joback Method
dvisc	0.0000125	Paxs	676.46	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R326368&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R326368&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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