

# 3-Buten-1-ol, propanoate

<b>Other names:</b>	3-Butenyl propionate Propanoic acid, 3-butenyl ester 3-Butenyl propanoate
<b>Inchi:</b>	InChI=1S/C7H12O2/c1-3-5-6-9-7(8)4-2/h3H,1,4-6H2,2H3
<b>InchiKey:</b>	IVOGAUVYWHQIBD-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O2
<b>SMILES:</b>	C=CCCOC(=O)CC
<b>Mol. weight [g/mol]:</b>	128.17
<b>CAS:</b>	27819-06-3

## Physical Properties

Property code	Value	Unit	Source
gf	-138.02	kJ/mol	Joback Method
hf	-307.18	kJ/mol	Joback Method
hfus	15.39	kJ/mol	Joback Method
hvap	39.66	kJ/mol	Joback Method
log10ws	-1.47		Crippen Method
logp	1.516		Crippen Method
mcvol	112.630	ml/mol	McGowan Method
pc	3079.57	kPa	Joback Method
rinpol	889.00		NIST Webbook
rinpol	864.00		NIST Webbook
rinpol	855.00		NIST Webbook
rinpol	894.00		NIST Webbook
rinpol	877.00		NIST Webbook
ripol	1203.00		NIST Webbook
ripol	1200.00		NIST Webbook
ripol	1230.00		NIST Webbook
tb	432.53	K	Joback Method
tc	612.55	K	Joback Method
tf	239.05	K	Joback Method
vc	0.432	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	222.85	J/molxK	432.53	Joback Method
cpg	233.32	J/molxK	462.53	Joback Method
cpg	243.41	J/molxK	492.54	Joback Method
cpg	253.12	J/molxK	522.54	Joback Method
cpg	262.44	J/molxK	552.54	Joback Method
cpg	271.40	J/molxK	582.54	Joback Method
cpg	279.98	J/molxK	612.55	Joback Method
dvisc	0.0028577	Paxs	239.05	Joback Method
dvisc	0.0015177	Paxs	271.30	Joback Method
dvisc	0.0009221	Paxs	303.54	Joback Method
dvisc	0.0006165	Paxs	335.79	Joback Method
dvisc	0.0004423	Paxs	368.04	Joback Method
dvisc	0.0003347	Paxs	400.28	Joback Method
dvisc	0.0002641	Paxs	432.53	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C27819063&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C27819063&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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