

# Cyclobutaneethanol, 1-methyl-2-(1-methylethenyl)-, trans-

**Other names:** Cyclobutaneethanol, 2-isopropenyl-1-methyl-, stereoisomer

Fragranol

**Inchi:** InChI=1S/C10H18O/c1-8(2)9-4-5-10(9,3)6-7-11/h9,11H,1,4-7H2,2-3H3

**InchiKey:** SJKPJXGGNKMRPD-UHFFFAOYSA-N

**Formula:** C10H18O

**SMILES:** C=C(C)C1CCC1(C)CCO

**Mol. weight [g/mol]:** 154.25

**CAS:** 30346-21-5

## Physical Properties

Property code	Value	Unit	Source
gf	11.24	kJ/mol	Joback Method
hf	-224.78	kJ/mol	Joback Method
hfus	13.96	kJ/mol	Joback Method
hvap	52.57	kJ/mol	Joback Method
log10ws	-2.54		Crippen Method
logp	2.361		Crippen Method
mcvol	142.470	ml/mol	McGowan Method
pc	2853.57	kPa	Joback Method
rinpol	1221.60		NIST Webbook
rinpol	1213.00		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1212.00		NIST Webbook
rinpol	1196.00		NIST Webbook
ripol	1769.00		NIST Webbook
tb	523.52	K	Joback Method
tc	710.61	K	Joback Method
tf	281.64	K	Joback Method
vc	0.542	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	348.60	J/mol×K	523.52	Joback Method

cpg	363.16	J/mol×K	554.70	Joback Method
cpg	376.87	J/mol×K	585.88	Joback Method
cpg	389.82	J/mol×K	617.07	Joback Method
cpg	402.10	J/mol×K	648.25	Joback Method
cpg	413.80	J/mol×K	679.43	Joback Method
cpg	425.00	J/mol×K	710.61	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C30346215&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C30346215&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/12-269-0/Cyclobutaneethanol-1-methyl-2-1-methylethenyl-trans.pdf>

Generated by Cheméo on 2024-04-19 19:38:22.195314886 +0000 UTC m=+15844751.115892249.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.