

# 3-Hexanone, 2-methyl-

<b>Other names:</b>	2-METHYL-3-HEXANONE 2-Methylhexan-3-one Isopropyl propyl ketone Propyl isopropyl ketone
<b>Inchi:</b>	InChI=1S/C7H14O/c1-4-5-7(8)6(2)3/h6H,4-5H2,1-3H3
<b>InchiKey:</b>	HIGGFWFRAWSMBR-UHFFFAOYSA-N
<b>Formula:</b>	C7H14O
<b>SMILES:</b>	CCCC(=O)C(C)C
<b>Mol. weight [g/mol]:</b>	114.19
<b>CAS:</b>	7379-12-6

## Physical Properties

Property code	Value	Unit	Source
gf	-123.30	kJ/mol	Joback Method
hf	-305.67	kJ/mol	Joback Method
hfus	11.96	kJ/mol	Joback Method
hvap	37.53	kJ/mol	Joback Method
log10ws	-1.79		Crippen Method
logp	2.012		Crippen Method
mcvol	111.060	ml/mol	McGowan Method
pc	3022.28	kPa	Joback Method
rinpol	820.00		NIST Webbook
rinpol	819.00		NIST Webbook
rinpol	819.00		NIST Webbook
rinpol	820.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	820.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	820.00		NIST Webbook
tb	412.99	K	Joback Method
tc	593.63	K	Joback Method
tf	203.58	K	Joback Method
vc	0.427	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.88	J/molxK	593.63	Joback Method
cpg	269.42	J/molxK	563.52	Joback Method
cpg	259.52	J/molxK	533.42	Joback Method
cpg	249.19	J/molxK	503.31	Joback Method
cpg	238.41	J/molxK	473.20	Joback Method
cpg	227.17	J/molxK	443.10	Joback Method
cpg	215.47	J/molxK	412.99	Joback Method
dvisc	0.0065747	Paxs	203.58	Joback Method
dvisc	0.0002897	Paxs	412.99	Joback Method
dvisc	0.0003833	Paxs	378.09	Joback Method
dvisc	0.0005371	Paxs	343.19	Joback Method
dvisc	0.0008121	Paxs	308.28	Joback Method
dvisc	0.0013648	Paxs	273.38	Joback Method
dvisc	0.0026700	Paxs	238.48	Joback Method
hvapt	41.30	kJ/mol	351.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49031e+01
Coeff. B	-3.62991e+03
Coeff. C	-5.40600e+01
Temperature range (K), min.	302.42
Temperature range (K), max.	432.51

## Sources

<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol1208.mol">https://www.thermo.com/files/research/kdb/mol/mol1208.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7379126&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7379126&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

**cpg:** Ideal gas heat capacity  
**dvisc:** Dynamic viscosity  
**gf:** Standard Gibbs free energy of formation  
**hf:** Enthalpy of formation at standard conditions  
**hfus:** Enthalpy of fusion at standard conditions  
**hvap:** Enthalpy of vaporization at standard conditions  
**hvapt:** Enthalpy of vaporization at a given temperature  
**log10ws:** Log10 of Water solubility in mol/l  
**logp:** Octanol/Water partition coefficient  
**mcvol:** McGowan's characteristic volume  
**pc:** Critical Pressure  
**pvap:** Vapor pressure  
**rinpol:** Non-polar retention indices  
**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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