

# 2-Azacyclooctanone

<b>Other names:</b>	Hexahydro-2(1H)-azocinone 2-Ketoheptamethyleneimine «omega»-Heptanolactam 2(1H)-Azocinone, hexahydro- 7-Aminoheptanoic acid lactam Azacyclooctan-2-one Enantholactam «omega»-Enantholactam «zeta»-Enantholactam 7-Heptanelactam Octahydroazocin-2-one Oenantholactam 2-Perhydroazocinone Suberonisoxim 2H-Azocin-2-one, hexahydro- NSC 77088 hexahydroazocin-2(1H)-one
<b>Inchi:</b>	InChI=1S/C7H13NO/c9-7-5-3-1-2-4-6-8-7/h1-6H2,(H,8,9)
<b>InchiKey:</b>	CJYXCQLOZNMIFP-UHFFFAOYSA-N
<b>Formula:</b>	C7H13NO
<b>SMILES:</b>	O=C1CCCCCN1
<b>Mol. weight [g/mol]:</b>	127.18
<b>CAS:</b>	673-66-5

## Physical Properties

Property code	Value	Unit	Source
chs	-4263.90 ± 1.30	kJ/mol	NIST Webbook
gf	-18.86	kJ/mol	Joback Method
hf	-225.36	kJ/mol	Joback Method
hfus	9.55	kJ/mol	Joback Method
hvap	43.26	kJ/mol	Joback Method
ie	9.19	eV	NIST Webbook
log10ws	-1.61		Crippen Method
logp	1.067		Crippen Method
mcvol	110.180	ml/mol	McGowan Method
pc	4205.63	kPa	Joback Method
tb	508.69	K	Joback Method

tc	757.46	K	Joback Method
tf	308.20 ± 0.50	K	NIST Webbook
tt	310.29 ± 0.02	K	NIST Webbook
vc	0.390	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	244.25	J/mol×K	508.69	Joback Method
cpg	262.25	J/mol×K	550.15	Joback Method
cpg	279.38	J/mol×K	591.61	Joback Method
cpg	295.60	J/mol×K	633.08	Joback Method
cpg	310.86	J/mol×K	674.54	Joback Method
cpg	325.13	J/mol×K	716.00	Joback Method
cpg	338.35	J/mol×K	757.46	Joback Method
hfust	13.78	kJ/mol	310.30	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	422.20	K	1.30	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C673665&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C673665&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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