

# 4-Quinolinol

<b>Other names:</b>	4-Hydroxyquinoline Kynurine quinolin-4-ol
<b>Inchi:</b>	InChI=1S/C9H7NO/c11-9-5-6-10-8-4-2-1-3-7(8)9/h1-6H,(H,10,11)
<b>InchiKey:</b>	PMZDQRJGMBQBF-UHFFFAOYSA-N
<b>Formula:</b>	C9H7NO
<b>SMILES:</b>	Oc1ccnc2ccccc12
<b>Mol. weight [g/mol]:</b>	145.16
<b>CAS:</b>	611-36-9

## Physical Properties

Property code	Value	Unit	Source
chs	-4427.70 ± 1.60	kJ/mol	NIST Webbook
hf	20.80 ± 2.30	kJ/mol	NIST Webbook
hfs	-114.30 ± 2.00	kJ/mol	NIST Webbook
hsub	135.10 ± 1.10	kJ/mol	NIST Webbook
hsub	135.10 ± 1.10	kJ/mol	NIST Webbook
ie	8.20	eV	NIST Webbook
log10ws	-2.59		Crippen Method
logp	1.940		Crippen Method
mcpvol	110.300	ml/mol	McGowan Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hsubt	128.80 ± 1.10	kJ/mol	424.00	NIST Webbook

## Sources

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C611369&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume

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