

# 1,2:4,6-Di-O-isopropylidene-L-sorbopyranose

<b>Other names:</b>	L-Sorbose, bis-O-(1-methylethylidene)- Diacetone sorbose
<b>Inchi:</b>	InChI=1S/C12H20O6/c1-10(2)15-6-12(18-10)9-8(13)7(5-14-12)16-11(3,4)17-9/h7-9,13H,
<b>InchiKey:</b>	HIHGMQOBMJOZKO-UHFFFAOYSA-N
<b>Formula:</b>	C12H20O6
<b>SMILES:</b>	CC1(C)OC2COC3(COC(C)(C)O3)C(O1)C2O
<b>Mol. weight [g/mol]:</b>	260.28
<b>CAS:</b>	32717-65-0

## Physical Properties

Property code	Value	Unit	Source
gf	-423.01	kJ/mol	Joback Method
hf	-924.78	kJ/mol	Joback Method
hfus	41.14	kJ/mol	Joback Method
hvap	77.58	kJ/mol	Joback Method
log10ws	-1.39		Crippen Method
logp	0.377		Crippen Method
mcvol	182.580	ml/mol	McGowan Method
pc	3096.73	kPa	Joback Method
rinpol	1571.00		NIST Webbook
rinpol	1534.00		NIST Webbook
tb	724.90	K	Joback Method
tc	951.13	K	Joback Method
tf	517.39	K	Joback Method
vc	0.661	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	602.64	J/molxK	724.90	Joback Method
cpg	620.05	J/molxK	762.61	Joback Method
cpg	637.43	J/molxK	800.31	Joback Method
cpg	655.11	J/molxK	838.02	Joback Method
cpg	673.46	J/molxK	875.72	Joback Method

cpg	692.82	J/mol×K	913.43	Joback Method
cpg	713.54	J/mol×K	951.13	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C32717650&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C32717650&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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