

# 2-Thiatricyclo[3.3.1.1(3,7)]decane

<b>Other names:</b>	2-Thiaadamantane
<b>Inchi:</b>	InChI=1S/C9H14S/c1-6-2-8-4-7(1)5-9(3-6)10-8/h6-9H,1-5H2
<b>InchiKey:</b>	NOBCRSKNBOQNEL-UHFFFAOYSA-N
<b>Formula:</b>	C9H14S
<b>SMILES:</b>	C1C2CC3CC1CC(C2)S3
<b>Mol. weight [g/mol]:</b>	154.27
<b>CAS:</b>	281-25-4

## Physical Properties

Property code	Value	Unit	Source
chs	-6000.86 ± 0.96	kJ/mol	NIST Webbook
gf	227.20	kJ/mol	Joback Method
hf	8.07	kJ/mol	Joback Method
hfs	-143.50 ± 1.10	kJ/mol	NIST Webbook
hfus	16.10	kJ/mol	Joback Method
hvap	41.04	kJ/mol	Joback Method
log10ws	-2.90		Crippen Method
logp	2.680		Crippen Method
mcvol	121.440	ml/mol	McGowan Method
pc	3435.91	kPa	Joback Method
rinpol	1439.00		NIST Webbook
rinpol	1429.00		NIST Webbook
rinpol	1449.00		NIST Webbook
tb	472.97	K	Joback Method
tc	704.84	K	Joback Method
tf	597.00 ± 3.00	K	NIST Webbook
vc	0.448	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	279.72	J/mol×K	472.97	Joback Method
cpg	299.56	J/mol×K	511.61	Joback Method
cpg	317.85	J/mol×K	550.26	Joback Method

cpg	334.73	J/mol×K	588.90	Joback Method
cpg	350.31	J/mol×K	627.55	Joback Method
cpg	364.71	J/mol×K	666.19	Joback Method
cpg	378.05	J/mol×K	704.84	Joback Method
cps	213.00	J/mol×K	298.15	NIST Webbook
hfust	8.10	kJ/mol	597.00	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C281254&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C281254&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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