

# 1-(2-propenyl)-trans-2-pentyl-cyclopropane

<b>Inchi:</b>	InChI=1S/C11H20/c1-3-5-6-8-11-9-10(11)7-4-2/h4,10-11H,2-3,5-9H2,1H3/t10-,11-/m1/s1
<b>InchiKey:</b>	FVVIJURMZEGKS-GHMZBOCLSA-N
<b>Formula:</b>	C11H20
<b>SMILES:</b>	C=CCC1CC1CCCC
<b>Mol. weight [g/mol]:</b>	152.28

## Physical Properties

Property code	Value	Unit	Source
gf	182.62	kJ/mol	Joback Method
hf	-92.48	kJ/mol	Joback Method
hfus	22.17	kJ/mol	Joback Method
hvap	39.01	kJ/mol	Joback Method
log10ws	-3.69		Crippen Method
logp	3.779		Crippen Method
mvol	150.690	ml/mol	McGowan Method
pc	2193.84	kPa	Joback Method
rinpol	1039.70		NIST Webbook
rinpol	1038.00		NIST Webbook
tb	449.83	K	Joback Method
tc	627.04	K	Joback Method
tf	225.67	K	Joback Method
vc	0.589	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	329.05	J/molxK	449.83	Joback Method
cpg	406.91	J/molxK	597.50	Joback Method
cpg	392.82	J/molxK	567.97	Joback Method
cpg	378.03	J/molxK	538.43	Joback Method
cpg	362.49	J/molxK	508.90	Joback Method
cpg	346.17	J/molxK	479.36	Joback Method
cpg	420.32	J/molxK	627.04	Joback Method
dvisc	0.0004029	Paxs	449.83	Joback Method

dvisc	0.0004472	Paxs	412.47	Joback Method
dvisc	0.0005067	Paxs	375.11	Joback Method
dvisc	0.0005902	Paxs	337.75	Joback Method
dvisc	0.0007141	Paxs	300.39	Joback Method
dvisc	0.0009121	Paxs	263.03	Joback Method
dvisc	0.0012633	Paxs	225.67	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R137049&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R137049&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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