

# Oxirane, 2,2'-[1,4-butanediylbis(oxymethylene)]bis-

<b>Other names:</b>	Butane, 1,4-bis(2,3-epoxypropoxy)- [Tetramethylenebis(oxymethylene)]dioxirane Butanediol diglycidyl ether Tetramethylene glycol diglycidyl ether 1,4-Bis(glycidyoxy)butane 1,4-Bis(2,3-epoxypropoxy)butane 1,4-Bis(2,3-epoxypropyloxy)butane 1,4-Butane diglycidyl ether 1,4-Butanediol diglycidyl ether 1,4-Diglycidyoxybutane 2,2'-[1,4-Butanediylbis(oxymethylene)]bis[oxirane] 1,4-Bis(oxiranylmethoxy)butane Araldit dy 026 Butane-1:4-diol diglycidyl ether CD 15006 A ChS-RR2 1,4-Diglycidloxybutane Grilonit RV 1806 TK 10352 NSC 24834
<b>Inchi:</b>	InChI=1S/C10H18O4/c1(3-11-5-9-7-13-9)2-4-12-6-10-8-14-10/h9-10H,1-8H2
<b>InchiKey:</b>	SHKUQQIDMUMQQK-UHFFFAOYSA-N
<b>Formula:</b>	C10H18O4
<b>SMILES:</b>	C(CCOCC1CO1)COCC1CO1
<b>Mol. weight [g/mol]:</b>	202.25
<b>CAS:</b>	2425-79-8

## Physical Properties

Property code	Value	Unit	Source
gf	-227.42	kJ/mol	Joback Method
hf	-632.57	kJ/mol	Joback Method
hfus	36.26	kJ/mol	Joback Method
hvap	51.52	kJ/mol	Joback Method
log10ws	-0.37		Crippen Method
logp	0.597		Crippen Method
mvol	153.520	ml/mol	McGowan Method
pc	2576.72	kPa	Joback Method

tb	540.42	K	Joback Method
tc	727.26	K	Joback Method
tf	335.94	K	Joback Method
vc	0.588	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	406.68	J/mol×K	540.42	Joback Method
cpg	422.22	J/mol×K	571.56	Joback Method
cpg	436.99	J/mol×K	602.70	Joback Method
cpg	451.00	J/mol×K	633.84	Joback Method
cpg	464.30	J/mol×K	664.98	Joback Method
cpg	476.91	J/mol×K	696.12	Joback Method
cpg	488.87	J/mol×K	727.26	Joback Method
dvisc	0.0021874	Paxs	335.94	Joback Method
dvisc	0.0017006	Paxs	370.02	Joback Method
dvisc	0.0013794	Paxs	404.10	Joback Method
dvisc	0.0011560	Paxs	438.18	Joback Method
dvisc	0.0009937	Paxs	472.26	Joback Method
dvisc	0.0008718	Paxs	506.34	Joback Method
dvisc	0.0007776	Paxs	540.42	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	430.70	K	1.50	NIST Webbook

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2425798&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/12-593-0/Oxirane-2-2-1-4-butanediylbis-oxymethylene-bis.pdf>

Generated by Cheméo on 2024-04-19 21:01:39.176453117 +0000 UTC m=+15849748.097030432.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.