

# Methanone, diphenyl-, hydrazone

<b>Other names:</b>	Benzophenone, hydrazone Benzophenoneimine N-amino-
<b>Inchi:</b>	InChI=1S/C13H12N2/c14-15-13(11-7-3-1-4-8-11)12-9-5-2-6-10-12/h1-10H,14H2
<b>InchiKey:</b>	QYCSNMDOZNUZIT-UHFFFAOYSA-N
<b>Formula:</b>	C13H12N2
<b>SMILES:</b>	NN=C(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	196.25
<b>CAS:</b>	5350-57-2

## Physical Properties

Property code	Value	Unit	Source
hf	267.63	kJ/mol	Joback Method
hvap	63.12	kJ/mol	Joback Method
log10ws	-3.26		Crippen Method
logp	2.398		Crippen Method
mcvol	162.170	ml/mol	McGowan Method
pc	2931.34	kPa	Joback Method
tb	699.29	K	Joback Method
tc	970.62	K	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	500.70	K	7.30	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5350572&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5350572&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature

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