

# Atrolactic acid

<b>Other names:</b>	Benzeneacetic acid, «alpha»-hydroxy-«alpha»-methyl- Mandelic acid, «alpha»-methyl- «alpha»-Hydroxy-«alpha»-phenylpropionic acid «alpha»-Phenyllactic acid 2-Hydroxy-2-phenylpropionic acid 2-Phenyl-2-hydroxypropionic acid 2-Phenyllactic acid dl-Atrolactic acid 2-Hydroxy-2-phenylpropanoic acid DL-2-Phenyllactic acid DL-«alpha»-Phenyllactic acid NSC 128998 NSC 401846
<b>Inchi:</b>	InChI=1S/C9H10O3/c1-9(12,8(10)11)7-5-3-2-4-6-7/h2-6,12H,1H3,(H,10,11)
<b>InchiKey:</b>	NWCHELUCVWSRRS-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O3
<b>SMILES:</b>	CC(O)(C(=O)O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	166.17
<b>CAS:</b>	515-30-0

## Physical Properties

Property code	Value	Unit	Source
gf	-262.41	kJ/mol	Joback Method
hf	-418.35	kJ/mol	Joback Method
hfus	15.47	kJ/mol	Joback Method
hvap	76.71	kJ/mol	Joback Method
log10ws	-1.22		Crippen Method
logp	0.979		Crippen Method
mcvol	127.220	ml/mol	McGowan Method
pc	4534.68	kPa	Joback Method
tb	667.00	K	Joback Method
tc	868.13	K	Joback Method
tf	391.60	K	Joback Method
vc	0.465	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	329.29	J/molxK	667.00	Joback Method
cpg	366.90	J/molxK	834.61	Joback Method
cpg	360.46	J/molxK	801.09	Joback Method
cpg	353.53	J/molxK	767.56	Joback Method
cpg	346.05	J/molxK	734.04	Joback Method
cpg	337.99	J/molxK	700.52	Joback Method
cpg	372.89	J/molxK	868.13	Joback Method
dvisc	0.0000186	Paxs	667.00	Joback Method
dvisc	0.0000328	Paxs	621.10	Joback Method
dvisc	0.0000634	Paxs	575.20	Joback Method
dvisc	0.0001371	Paxs	529.30	Joback Method
dvisc	0.0003432	Paxs	483.40	Joback Method
dvisc	0.0010421	Paxs	437.50	Joback Method
dvisc	0.0041052	Paxs	391.60	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C515300&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C515300&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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