

# Tetrahydrofuran, 3-chloro-2-(2-chloroethoxy)

<b>Other names:</b>	2-(2-Chloro-ethoxy)-3-chloro-tetrahydro-furan
<b>Inchi:</b>	InChI=1S/C6H10Cl2O2/c7-2-4-10-6-5(8)1-3-9-6/h5-6H,1-4H2
<b>InchiKey:</b>	OIJQDJIQKWWGGW-UHFFFAOYSA-N
<b>Formula:</b>	C6H10Cl2O2
<b>SMILES:</b>	CICCOCC1OCCC1Cl
<b>Mol. weight [g/mol]:</b>	185.05

## Physical Properties

Property code	Value	Unit	Source
gf	-186.50	kJ/mol	Joback Method
hf	-422.73	kJ/mol	Joback Method
hfus	23.86	kJ/mol	Joback Method
hvap	44.59	kJ/mol	Joback Method
log10ws	-1.43		Crippen Method
logp	1.596		Crippen Method
mcvol	120.760	ml/mol	McGowan Method
pc	3276.53	kPa	Joback Method
rinpol	1210.00		NIST Webbook
rinpol	1210.00		NIST Webbook
tb	471.52	K	Joback Method
tc	680.98	K	Joback Method
tf	272.68	K	Joback Method
vc	0.449	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	244.99	J/mol×K	471.52	Joback Method
cpg	257.59	J/mol×K	506.43	Joback Method
cpg	269.58	J/mol×K	541.34	Joback Method
cpg	280.97	J/mol×K	576.25	Joback Method
cpg	291.75	J/mol×K	611.16	Joback Method
cpg	301.95	J/mol×K	646.07	Joback Method
cpg	311.56	J/mol×K	680.98	Joback Method

dvisc	0.0026616	Paxs	272.68	Joback Method
dvisc	0.0016324	Paxs	305.82	Joback Method
dvisc	0.0011016	Paxs	338.96	Joback Method
dvisc	0.0007973	Paxs	372.10	Joback Method
dvisc	0.0006084	Paxs	405.24	Joback Method
dvisc	0.0004837	Paxs	438.38	Joback Method
dvisc	0.0003971	Paxs	471.52	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R91226&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R91226&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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