

# 14-Heptacosanone

<b>Other names:</b>	Ditridecyl ketone heptacosan-14-one
<b>Inchi:</b>	InChI=1S/C27H54O/c1-3-5-7-9-11-13-15-17-19-21-23-25-27(28)26-24-22-20-18-16-14-1
<b>InchiKey:</b>	VCZMOZVQLARCOE-UHFFFAOYSA-N
<b>Formula:</b>	C27H54O
<b>SMILES:</b>	CCCCCCCCCCCCC(=O)CCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	394.72
<b>CAS:</b>	542-50-7

## Physical Properties

Property code	Value	Unit	Source
gf	47.54	kJ/mol	Joback Method
hf	-713.19	kJ/mol	Joback Method
hfus	67.28	kJ/mol	Joback Method
hvap	82.44	kJ/mol	Joback Method
log10ws	-10.40		Crippen Method
logp	9.958		Crippen Method
mvol	392.860	ml/mol	McGowan Method
pc	712.25	kPa	Joback Method
tb	871.03	K	Joback Method
tc	1068.20	K	Joback Method
tf	443.98	K	Joback Method
vc	1.554	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1315.68	J/molxK	871.03	Joback Method
cpg	1339.36	J/molxK	903.89	Joback Method
cpg	1361.73	J/molxK	936.75	Joback Method
cpg	1382.84	J/molxK	969.62	Joback Method
cpg	1402.77	J/molxK	1002.48	Joback Method
cpg	1421.57	J/molxK	1035.34	Joback Method
cpg	1439.30	J/molxK	1068.20	Joback Method

dvisc	0.0011050	Paxs	443.98	Joback Method
dvisc	0.0004213	Paxs	515.15	Joback Method
dvisc	0.0002030	Paxs	586.33	Joback Method
dvisc	0.0001145	Paxs	657.50	Joback Method
dvisc	0.0000723	Paxs	728.68	Joback Method
dvisc	0.0000495	Paxs	799.86	Joback Method
dvisc	0.0000361	Paxs	871.03	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.65724e+01
Coeff. B	-6.83660e+03
Coeff. C	-1.45105e+02
Temperature range (K), min.	564.92
Temperature range (K), max.	752.21

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C542507&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C542507&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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