

# 3-(Benzylmethylamino)-1-propanol

<b>Other names:</b>	1-Propanol, 3-[methyl(phenylmethyl)amino]- 3-(N-benzyl-N-methylamino)propan-1-ol
<b>Inchi:</b>	InChI=1S/C11H17NO/c1-12(8-5-9-13)10-11-6-3-2-4-7-11/h2-4,6-7,13H,5,8-10H2,1H3
<b>InchiKey:</b>	YQAUZGVTGQVMPM-UHFFFAOYSA-N
<b>Formula:</b>	C11H17NO
<b>SMILES:</b>	CN(CCCO)Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	179.26
<b>CAS:</b>	5814-42-6

## Physical Properties

Property code	Value	Unit	Source
gf	128.11	kJ/mol	Joback Method
hf	-118.54	kJ/mol	Joback Method
hfus	25.40	kJ/mol	Joback Method
hvap	61.08	kJ/mol	Joback Method
log10ws	-1.86		Crippen Method
logp	1.501		Crippen Method
mcvol	157.940	ml/mol	McGowan Method
pc	2912.39	kPa	Joback Method
tb	582.38	K	Joback Method
tc	770.30	K	Joback Method
tf	333.44	K	Joback Method
vc	0.581	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	394.44	J/molxK	582.38	Joback Method
cpg	408.18	J/molxK	613.70	Joback Method
cpg	421.14	J/molxK	645.02	Joback Method
cpg	433.37	J/molxK	676.34	Joback Method
cpg	444.89	J/molxK	707.66	Joback Method
cpg	455.75	J/molxK	738.98	Joback Method
cpg	465.97	J/molxK	770.30	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	406.70	K	0.50	NIST Webbook

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5814426&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5814426&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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