

# 1-nitro-3-methylbutane

<b>Other names:</b>	Butane, 3-methyl-1-nitro- Nitro-3-methylbutane 3-methyl-1-nitrobutane
<b>Inchi:</b>	InChI=1S/C5H11NO2/c1-5(2)3-4-6(7)8/h5H,3-4H2,1-2H3
<b>InchiKey:</b>	FEJLPMVSVDSKHJ-UHFFFAOYSA-N
<b>Formula:</b>	C5H11NO2
<b>SMILES:</b>	CC(C)CC[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	117.15
<b>CAS:</b>	627-67-8

## Physical Properties

Property code	Value	Unit	Source
gf	24.33	kJ/mol	Joback Method
hf	-162.57	kJ/mol	Joback Method
hfus	16.54	kJ/mol	Joback Method
hvap	42.93	kJ/mol	Joback Method
log10ws	-1.76		Crippen Method
logp	1.309		Crippen Method
mcvol	98.730	ml/mol	McGowan Method
pc	3551.53	kPa	Joback Method
ripol	856.00		NIST Webbook
ripol	887.00		NIST Webbook
ripol	1308.00		NIST Webbook
ripol	1339.00		NIST Webbook
ripol	1317.00		NIST Webbook
ripol	1308.00		NIST Webbook
tb	465.20	K	Joback Method
tc	674.41	K	Joback Method
tf	274.72	K	Joback Method
vc	0.392	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	209.31	J/mol×K	465.20	Joback Method
cpg	220.22	J/mol×K	500.07	Joback Method
cpg	230.58	J/mol×K	534.94	Joback Method
cpg	240.40	J/mol×K	569.81	Joback Method
cpg	249.70	J/mol×K	604.68	Joback Method
cpg	258.50	J/mol×K	639.54	Joback Method
cpg	266.81	J/mol×K	674.41	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C627678&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C627678&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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