

2-Methyl-3-buten-2-ol, heptafluorobutyrate

Inchi: InChI=1S/C9H9F7O2/c1-4-6(2,3)18-5(17)7(10,11)8(12,13)9(14,15)16/h4H,1H2,2-3H3
InchiKey: OACFSDYVIMZLIH-UHFFFAOYSA-N
Formula: C9H9F7O2
SMILES: C=CC(C)(C)OC(=O)C(F)(F)C(F)(F)C(F)(F)F
Mol. weight [g/mol]: 282.16

Physical Properties

Property code	Value	Unit	Source
gf	-1473.49	kJ/mol	Joback Method
hf	-1756.23	kJ/mol	Joback Method
hfus	12.48	kJ/mol	Joback Method
hvap	33.21	kJ/mol	Joback Method
log10ws	-3.71		Crippen Method
logp	3.327		Crippen Method
mcvol	153.200	ml/mol	McGowan Method
pc	2005.50	kPa	Joback Method
rinpol	833.10		NIST Webbook
rinpol	833.10		NIST Webbook
tb	460.26	K	Joback Method
tc	619.07	K	Joback Method
tf	275.40	K	Joback Method
vc	0.626	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	373.79	J/mol×K	460.26	Joback Method
cpg	386.68	J/mol×K	486.73	Joback Method
cpg	398.72	J/mol×K	513.20	Joback Method
cpg	409.94	J/mol×K	539.66	Joback Method
cpg	420.39	J/mol×K	566.13	Joback Method
cpg	430.10	J/mol×K	592.60	Joback Method
cpg	439.13	J/mol×K	619.07	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U352696&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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