

# 2-Octyne

<b>Other names:</b>	n-C <sub>5</sub> H <sub>11</sub> C«equiv»CCH <sub>3</sub> n-C <sub>5</sub> H <sub>11</sub> CÂ«equivÂ»CCH <sub>3</sub> oct-2-yne
<b>Inchi:</b>	InChI=1S/C <sub>8</sub> H <sub>14</sub> /c1-3-5-7-8-6-4-2/h3,5,7-8H <sub>2</sub> ,1-2H <sub>3</sub>
<b>InchiKey:</b>	QCQALVMFTWRCFI-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>8</sub> H <sub>14</sub>
<b>SMILES:</b>	CC#CCCCCC
<b>Mol. weight [g/mol]:</b>	110.20
<b>CAS:</b>	2809-67-8

## Physical Properties

Property code	Value	Unit	Source
gf	219.28	kJ/mol	Joback Method
hf	63.80 ± 1.50	kJ/mol	NIST Webbook
hfus	19.60	kJ/mol	Joback Method
hvap	44.52	kJ/mol	NIST Webbook
hvap	44.50 ± 0.10	kJ/mol	NIST Webbook
ie	9.30 ± 0.01	eV	NIST Webbook
ie	9.31 ± 0.01	eV	NIST Webbook
ie	9.31 ± 0.02	eV	NIST Webbook
log10ws	-2.97		Crippen Method
logp	2.590		Crippen Method
mcvol	114.980	ml/mol	McGowan Method
pc	2999.15	kPa	Joback Method
rinpol	842.00		NIST Webbook
rinpol	865.00		NIST Webbook
rinpol	860.00		NIST Webbook
rinpol	870.00		NIST Webbook
rinpol	870.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	863.00		NIST Webbook
rinpol	863.00		NIST Webbook
rinpol	862.00		NIST Webbook
rinpol	865.00		NIST Webbook
rinpol	861.00		NIST Webbook
rinpol	863.00		NIST Webbook

rinpol	864.00		NIST Webbook
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rinpol	863.00		NIST Webbook
rinpol	864.00		NIST Webbook
rinpol	841.00		NIST Webbook
rinpol	864.00		NIST Webbook
rinpol	865.00		NIST Webbook
rinpol	866.00		NIST Webbook
rinpol	864.00		NIST Webbook
rinpol	863.00		NIST Webbook
rinpol	862.00		NIST Webbook
rinpol	843.45		NIST Webbook
rinpol	843.46		NIST Webbook
rinpol	843.60		NIST Webbook
rinpol	854.00		NIST Webbook
rinpol	842.00		NIST Webbook
rinpol	841.00		NIST Webbook
rinpol	858.90		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	842.00		NIST Webbook
rinpol	859.00		NIST Webbook
rinpol	860.00		NIST Webbook
rinpol	860.00		NIST Webbook
rinpol	864.00		NIST Webbook
rinpol	845.20		NIST Webbook
rinpol	842.00		NIST Webbook
rinpol	845.00		NIST Webbook
rinpol	841.00		NIST Webbook
rinpol	840.00		NIST Webbook
rinpol	843.45		NIST Webbook
ripol	1041.00		NIST Webbook
ripol	1065.00		NIST Webbook
ripol	1063.00		NIST Webbook
ripol	1064.00		NIST Webbook
ripol	1054.00		NIST Webbook
ripol	1037.00		NIST Webbook
ripol	1079.00		NIST Webbook
ripol	1040.00		NIST Webbook
ripol	1037.00		NIST Webbook
ripol	1059.00		NIST Webbook
ripol	1079.00		NIST Webbook
tb	410.35 ± 0.80	K	NIST Webbook
tb	411.15 ± 0.20	K	NIST Webbook

tb	410.88 ± 0.50	K	NIST Webbook
tb	407.15 ± 2.00	K	NIST Webbook
tb	410.70	K	NIST Webbook
tb	411.20	K	NIST Webbook
tb	411.00	K	NIST Webbook
tb	411.35 ± 0.50	K	NIST Webbook
tb	409.15 ± 1.50	K	NIST Webbook
tb	409.40 ± 1.50	K	NIST Webbook
tb	409.65 ± 1.50	K	NIST Webbook
tb	410.25 ± 0.50	K	NIST Webbook
tb	411.24 ± 0.60	K	NIST Webbook
tc	578.71	K	Joback Method
tf	211.55 ± 0.20	K	NIST Webbook
tf	211.15 ± 1.50	K	NIST Webbook
tf	211.44 ± 2.00	K	NIST Webbook
vc	0.446	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	208.39	J/mol×K	391.44	Joback Method
cpg	220.42	J/mol×K	422.65	Joback Method
cpg	231.99	J/mol×K	453.86	Joback Method
cpg	243.10	J/mol×K	485.07	Joback Method
cpg	253.76	J/mol×K	516.28	Joback Method
cpg	264.00	J/mol×K	547.49	Joback Method
cpg	273.81	J/mol×K	578.71	Joback Method
hvapt	39.90	kJ/mol	390.00	NIST Webbook
hvapt	37.26	kJ/mol	410.70	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.39125e+01
Coeff. B	-3.18531e+03
Coeff. C	-6.81600e+01

Temperature range (K), min.	301.95
Temperature range (K), max.	438.50

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.71593e+01
Coeff. B	-8.11149e+03
Coeff. C	-1.06258e+01
Coeff. D	6.73236e-06
Temperature range (K), min.	368.15
Temperature range (K), max.	412.00

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermopedia.com/doc/thermoprop/kdb/mol/mol429.mol">https://www.thermopedia.com/doc/thermoprop/kdb/mol/mol429.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2809678&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2809678&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermopedia.com/doc/thermoprop/kdb/hcprop/showprop.php?cmpid=429">https://www.thermopedia.com/doc/thermoprop/kdb/hcprop/showprop.php?cmpid=429</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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<https://www.chemeo.com/cid/12-945-9/2-Octyne.pdf>

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