

Dichloroacetamide, N,N-dihexyl-

Inchi:	InChI=1S/C14H27Cl2NO/c1-3-5-7-9-11-17(14(18)13(15)16)12-10-8-6-4-2/h13H,3-12H2,
InchiKey:	TXPVLEZUMVIIBX-UHFFFAOYSA-N
Formula:	C14H27Cl2NO
SMILES:	CCCCCN(CCCCCC)C(=O)C(Cl)Cl
Mol. weight [g/mol]:	296.28

Physical Properties

Property code	Value	Unit	Source
gf	22.56	kJ/mol	Joback Method
hf	-414.10	kJ/mol	Joback Method
hfus	41.51	kJ/mol	Joback Method
hvap	63.93	kJ/mol	Joback Method
log10ws	-4.94		Crippen Method
logp	4.779		Crippen Method
mvol	244.150	ml/mol	McGowan Method
pc	1517.57	kPa	Joback Method
rinpol	1890.00		NIST Webbook
tb	660.45	K	Joback Method
tc	839.33	K	Joback Method
tf	374.78	K	Joback Method
vc	0.935	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	644.09	J/mol×K	660.45	Joback Method
cpg	660.24	J/mol×K	690.26	Joback Method
cpg	675.58	J/mol×K	720.08	Joback Method
cpg	690.13	J/mol×K	749.89	Joback Method
cpg	703.94	J/mol×K	779.70	Joback Method
cpg	717.02	J/mol×K	809.52	Joback Method
cpg	729.41	J/mol×K	839.33	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308634&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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