

# Succinic acid, 2,3-dichlorophenyl 4-chlorobenzyl ester

<b>Inchi:</b>	InChI=1S/C17H13Cl3O4/c18-12-6-4-11(5-7-12)10-23-15(21)8-9-16(22)24-14-3-1-2-13(19)
<b>InchiKey:</b>	RCHRFPNYADUPAD-UHFFFAOYSA-N
<b>Formula:</b>	C17H13Cl3O4
<b>SMILES:</b>	O=C(CCC(=O)Oc1cccc(Cl)c1Cl)OCc1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	387.64

## Physical Properties

Property code	Value	Unit	Source
gf	-215.44	kJ/mol	Joback Method
hf	-492.38	kJ/mol	Joback Method
hfus	44.87	kJ/mol	Joback Method
hvap	91.44	kJ/mol	Joback Method
log10ws	-6.07		Crippen Method
logp	5.076		Crippen Method
mvol	254.470	ml/mol	McGowan Method
pc	1971.80	kPa	Joback Method
rinpol	2968.00		NIST Webbook
rinpol	2968.00		NIST Webbook
tb	921.53	K	Joback Method
tc	1162.77	K	Joback Method
tf	605.83	K	Joback Method
vc	0.967	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	672.67	J/molxK	921.53	Joback Method
cpg	708.30	J/molxK	1122.56	Joback Method
cpg	703.48	J/molxK	1082.35	Joback Method
cpg	697.53	J/molxK	1042.15	Joback Method
cpg	690.43	J/molxK	1001.94	Joback Method
cpg	682.15	J/molxK	961.74	Joback Method
cpg	712.03	J/molxK	1162.77	Joback Method
dvisc	0.0000571	Paxs	921.53	Joback Method

dvisc	0.0000699	Paxs	868.91	Joback Method
dvisc	0.0000878	Paxs	816.30	Joback Method
dvisc	0.0001138	Paxs	763.68	Joback Method
dvisc	0.0001534	Paxs	711.06	Joback Method
dvisc	0.0002167	Paxs	658.45	Joback Method
dvisc	0.0003250	Paxs	605.83	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389681&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389681&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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