

# Pentafluorobenzoic acid, 2-butyl ester

<b>Other names:</b>	1-methylpropyl pentafluorobenzoate
<b>Inchi:</b>	InChI=1S/C11H9F5O2/c1-3-4(2)18-11(17)5-6(12)8(14)10(16)9(15)7(5)13/h4H,3H2,1-2H3
<b>InchiKey:</b>	NYBFFHZPCUJMMG-UHFFFAOYSA-N
<b>Formula:</b>	C11H9F5O2
<b>SMILES:</b>	CCC(C)OC(=O)c1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	268.18
<b>CAS:</b>	99483-12-2

## Physical Properties

Property code	Value	Unit	Source
gf	-1104.41	kJ/mol	Joback Method
hf	-1321.82	kJ/mol	Joback Method
hfus	31.01	kJ/mol	Joback Method
hvap	50.35	kJ/mol	Joback Method
log10ws	-4.74		Crippen Method
logp	3.337		Crippen Method
mcvol	158.380	ml/mol	McGowan Method
pc	2086.93	kPa	Joback Method
rinpol	1170.00		NIST Webbook
rinpol	1171.00		NIST Webbook
rinpol	1170.00		NIST Webbook
rinpol	1177.00		NIST Webbook
rinpol	1165.00		NIST Webbook
rinpol	1171.00		NIST Webbook
ripol	1444.00		NIST Webbook
ripol	1408.00		NIST Webbook
ripol	1444.00		NIST Webbook
ripol	1424.00		NIST Webbook
ripol	1424.00		NIST Webbook
ripol	1421.00		NIST Webbook
tb	574.86	K	Joback Method
tc	749.37	K	Joback Method
tf	362.86	K	Joback Method
vc	0.651	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	385.86	J/mol×K	574.86	Joback Method
cpg	396.43	J/mol×K	603.95	Joback Method
cpg	406.57	J/mol×K	633.03	Joback Method
cpg	416.27	J/mol×K	662.12	Joback Method
cpg	425.54	J/mol×K	691.20	Joback Method
cpg	434.37	J/mol×K	720.29	Joback Method
cpg	442.76	J/mol×K	749.37	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C99483122&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C99483122&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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