

Glutaric acid, 3-chlorophenyl pentafluorophenyl ester

Inchi: InChI=1S/C17H10ClF5O4/c18-8-3-1-4-9(7-8)26-10(24)5-2-6-11(25)27-17-15(22)13(20)1
InchiKey: CRTXKBDYKNRLNN-UHFFFAOYSA-N
Formula: C17H10ClF5O4
SMILES: O=C(CCCC(=O)Oc1c(F)c(F)c(F)c(F)c1F)Oc1cccc(Cl)c1
Mol. weight [g/mol]: 408.70

Physical Properties

Property code	Value	Unit	Source
gf	-1194.52	kJ/mol	Joback Method
hf	-1475.86	kJ/mol	Joback Method
hfus	50.70	kJ/mol	Joback Method
hvap	80.57	kJ/mol	Joback Method
log10ws	-6.51		Crippen Method
logp	4.717		Crippen Method
mvol	238.840	ml/mol	McGowan Method
pc	1668.70	kPa	Joback Method
rinpol	2232.00		NIST Webbook
rinpol	2232.00		NIST Webbook
tb	857.96	K	Joback Method
tc	1064.21	K	Joback Method
tf	586.50	K	Joback Method
vc	0.959	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	665.41	J/mol×K	857.96	Joback Method
cpg	675.19	J/mol×K	892.33	Joback Method
cpg	684.03	J/mol×K	926.71	Joback Method
cpg	691.93	J/mol×K	961.08	Joback Method
cpg	698.89	J/mol×K	995.46	Joback Method
cpg	704.91	J/mol×K	1029.83	Joback Method
cpg	709.99	J/mol×K	1064.21	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U392112&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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