

2,2,3-Trichloro-n-(4-chloro-2-methylphenyl)propanamide

Inchi:	InChI=1S/C10H9Cl4NO/c1-6-4-7(12)2-3-8(6)15-9(16)10(13,14)5-11/h2-4H,5H2,1H3,(H,1)
InchiKey:	YVXKSFOYTJUMDP-UHFFFAOYSA-N
Formula:	C10H9Cl4NO
SMILES:	<chem>Cc1cc(Cl)ccc1NC(=O)C(Cl)(Cl)CCl</chem>
Mol. weight [g/mol]:	301.00
CAS:	116402-46-1

Physical Properties

Property code	Value	Unit	Source
gf	42.06	kJ/mol	Joback Method
hf	-166.96	kJ/mol	Joback Method
hfus	30.99	kJ/mol	Joback Method
hvap	70.88	kJ/mol	Joback Method
log10ws	-4.32		Crippen Method
logp	4.000		Crippen Method
mcvol	188.510	ml/mol	McGowan Method
pc	2684.64	kPa	Joback Method
tb	715.37	K	Joback Method
tc	958.74	K	Joback Method
tf	478.61	K	Joback Method
vc	0.714	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	418.49	J/molxK	715.37	Joback Method
cpg	428.20	J/molxK	755.93	Joback Method
cpg	437.05	J/molxK	796.49	Joback Method
cpg	445.11	J/molxK	837.05	Joback Method
cpg	452.45	J/molxK	877.62	Joback Method
cpg	459.17	J/molxK	918.18	Joback Method
cpg	465.32	J/molxK	958.74	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C116402461&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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