

# Isophthalic acid, isobutyl pentachlorophenyl ester

Inchi:	InChI=1S/C18H13Cl5O4/c1-8(2)7-26-17(24)9-4-3-5-10(6-9)18(25)27-16-14(22)12(20)11
InchiKey:	LSEIWOOCCOPTAH-UHFFFAOYSA-N
Formula:	C18H13Cl5O4
SMILES:	CC(C)COC(=O)c1cccc(C(=O)Oc2c(Cl)c(Cl)c(Cl)c(Cl)c2Cl)c1
Mol. weight [g/mol]:	470.56

## Physical Properties

Property code	Value	Unit	Source
gf	-262.21	kJ/mol	Joback Method
hf	-584.19	kJ/mol	Joback Method
hfus	51.16	kJ/mol	Joback Method
hvap	104.04	kJ/mol	Joback Method
log10ws	-8.24		Crippen Method
logp	6.986		Crippen Method
mvol	293.040	ml/mol	McGowan Method
pc	1653.80	kPa	Joback Method
rinpol	3167.00		NIST Webbook
rinpol	3167.00		NIST Webbook
tb	1033.77	K	Joback Method
tc	1284.55	K	Joback Method
tf	699.50	K	Joback Method
vc	1.115	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	755.84	J/molxK	1033.77	Joback Method
cpg	762.31	J/molxK	1075.57	Joback Method
cpg	767.41	J/molxK	1117.36	Joback Method
cpg	771.15	J/molxK	1159.16	Joback Method
cpg	773.54	J/molxK	1200.96	Joback Method
cpg	774.59	J/molxK	1242.75	Joback Method
cpg	774.31	J/molxK	1284.55	Joback Method
dvisc	0.0001670	Paxs	699.50	Joback Method

dvisc	0.0001167	Paxs	755.21	Joback Method
dvisc	0.0000857	Paxs	810.92	Joback Method
dvisc	0.0000655	Paxs	866.63	Joback Method
dvisc	0.0000517	Paxs	922.35	Joback Method
dvisc	0.0000419	Paxs	978.06	Joback Method
dvisc	0.0000347	Paxs	1033.77	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U356575&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U356575&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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