

# Tetramethylene diisocyanate

<b>Other names:</b>	Butane, 1,4-diisocyanato- 1,4-Diisocyanatobutane
<b>Inchi:</b>	InChI=1S/C6H8N2O2/c9-5-7-3-1-2-4-8-6-10/h1-4H2
<b>InchiKey:</b>	OVBFMUAFNIIQAL-UHFFFAOYSA-N
<b>Formula:</b>	C6H8N2O2
<b>SMILES:</b>	O=C=NCCCCN=C=O
<b>Mol. weight [g/mol]:</b>	140.14
<b>CAS:</b>	4538-37-8

## Physical Properties

Property code	Value	Unit	Source
hf	-177.99	kJ/mol	Joback Method
hvap	48.01	kJ/mol	Joback Method
log10ws	-9.33		Crippen Method
logp	0.438		Crippen Method
mcvol	109.900	ml/mol	McGowan Method
pc	3690.97	kPa	Joback Method
tb	470.02	K	Joback Method
tc	659.16	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hfust	20.76	kJ/mol	184.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4538378&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4538378&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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