

# Phthalic acid, 3,5-dinitro-4-methylbenzyl propyl ester

<b>Inchi:</b>	InChI=1S/C19H18N2O8/c1-3-8-28-18(22)14-6-4-5-7-15(14)19(23)29-11-13-9-16(20(24)2
<b>InchiKey:</b>	ZUNYYKSSZDXARE-UHFFFAOYSA-N
<b>Formula:</b>	C19H18N2O8
<b>SMILES:</b>	CCCOC(=O)c1ccccc1C(=O)OCc1cc([N+](=O)[O-])c(C)c([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	402.35

## Physical Properties

Property code	Value	Unit	Source
gf	-101.34	kJ/mol	Joback Method
hf	-519.43	kJ/mol	Joback Method
hfus	59.79	kJ/mol	Joback Method
hvap	116.58	kJ/mol	Joback Method
log10ws	-6.68		Crippen Method
logp	3.735		Crippen Method
mcvol	280.770	ml/mol	McGowan Method
pc	1851.52	kPa	Joback Method
rinpol	3079.00		NIST Webbook
rinpol	3079.00		NIST Webbook
tb	1163.66	K	Joback Method
tc	1432.19	K	Joback Method
tf	838.35	K	Joback Method
vc	1.095	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	888.42	J/mol×K	1163.66	Joback Method
cpg	892.70	J/mol×K	1208.42	Joback Method
cpg	895.31	J/mol×K	1253.17	Joback Method
cpg	896.28	J/mol×K	1297.93	Joback Method
cpg	895.66	J/mol×K	1342.68	Joback Method
cpg	893.50	J/mol×K	1387.44	Joback Method
cpg	889.83	J/mol×K	1432.19	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415510&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415510&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvpap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinppl:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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