

# Glutaric acid, 2,4-dichloronaphthyl hexyl ester

<b>Inchi:</b>	InChI=1S/C21H24Cl2O4/c1-2-3-4-7-13-26-19(24)11-8-12-20(25)27-21-16-10-6-5-9-15(16)
<b>InchiKey:</b>	NDORMXGKKWXWQL-UHFFFAOYSA-N
<b>Formula:</b>	C21H24Cl2O4
<b>SMILES:</b>	CCCCCOC(=O)CCCC(=O)Oc1c(Cl)cc(Cl)c2cccc12
<b>Mol. weight [g/mol]:</b>	411.32

## Physical Properties

Property code	Value	Unit	Source
gf	-175.59	kJ/mol	Joback Method
hf	-604.66	kJ/mol	Joback Method
hfus	54.01	kJ/mol	Joback Method
hvap	95.32	kJ/mol	Joback Method
log10ws	-7.59		Crippen Method
logp	6.346		Crippen Method
mvol	302.890	ml/mol	McGowan Method
pc	1387.11	kPa	Joback Method
rinpol	3122.00		NIST Webbook
rinpol	3122.00		NIST Webbook
tb	967.92	K	Joback Method
tc	1193.03	K	Joback Method
tf	627.27	K	Joback Method
vc	1.171	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	905.28	J/molxK	967.92	Joback Method
cpg	956.67	J/molxK	1155.51	Joback Method
cpg	948.34	J/molxK	1117.99	Joback Method
cpg	939.08	J/molxK	1080.47	Joback Method
cpg	928.86	J/molxK	1042.96	Joback Method
cpg	917.61	J/molxK	1005.44	Joback Method
cpg	964.13	J/molxK	1193.03	Joback Method
dvisc	0.0000751	Paxs	967.92	Joback Method

dvisc	0.0000908	Paxs	911.14	Joback Method
dvisc	0.0001126	Paxs	854.37	Joback Method
dvisc	0.0001439	Paxs	797.60	Joback Method
dvisc	0.0001911	Paxs	740.82	Joback Method
dvisc	0.0002659	Paxs	684.04	Joback Method
dvisc	0.0003928	Paxs	627.27	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358926&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358926&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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