

# 1,4-Naphthalenedione, 2-amino-3-chloro-

<b>Other names:</b>	Quinoclamine ACN ACNQ 2-Amino-3-chloronaphthoquinone 2-Amino-3-chloro-1,4-naphthoquinone 2-Chloro-3-amino-1,4-naphthoquinone 3-Chloro-2-amino-1,4-naphthoquinone Mogeton 1,4-Naphthoquinone, 2-amino-3-chloro- 06K-Quinone O 6K-quinone Mogeton G Mogeton granule 06K 06K-50W NSC 3910 NSC 642009 Quinoclamine
<b>Inchi:</b>	InChI=1S/C10H6ClNO2/c11-7-8(12)10(14)6-4-2-1-3-5(6)9(7)13/h1-4H,12H2
<b>InchiKey:</b>	OBLNWSCLAYSJJR-UHFFFAOYSA-N
<b>Formula:</b>	C10H6ClNO2
<b>SMILES:</b>	<chem>NC1=C(Cl)C(=O)c2ccccc2C1=O</chem>
<b>Mol. weight [g/mol]:</b>	207.61
<b>CAS:</b>	2797-51-5

## Physical Properties

Property code	Value	Unit	Source
gf	12.50	kJ/mol	Joback Method
hf	-160.20	kJ/mol	Joback Method
hfus	19.13	kJ/mol	Joback Method
hvap	66.32	kJ/mol	Joback Method
log10ws	-3.12		Crippen Method
logp	1.475		Crippen Method
mcvol	138.200	ml/mol	McGowan Method
pc	4015.93	kPa	Joback Method
rinpol	1974.00		NIST Webbook
rinpol	1968.00		NIST Webbook

rinpol	1961.00		NIST Webbook
rinpol	1961.00		NIST Webbook
rinpol	1968.00		NIST Webbook
rinpol	1961.00		NIST Webbook
tb	730.26	K	Joback Method
tc	1005.50	K	Joback Method
tf	535.48	K	Joback Method
vc	0.515	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	342.29	J/mol×K	730.26	Joback Method
cpg	353.66	J/mol×K	776.13	Joback Method
cpg	363.99	J/mol×K	822.01	Joback Method
cpg	373.24	J/mol×K	867.88	Joback Method
cpg	381.39	J/mol×K	913.75	Joback Method
cpg	388.42	J/mol×K	959.63	Joback Method
cpg	394.29	J/mol×K	1005.50	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2797515&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2797515&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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