

# 6-Bromohexanoic acid, 3-ethylphenyl ester

<b>Inchi:</b>	InChI=1S/C14H19BrO2/c1-2-12-7-6-8-13(11-12)17-14(16)9-4-3-5-10-15/h6-8,11H,2-5,9-
<b>InchiKey:</b>	CGPARSXTTAMAJJ-UHFFFAOYSA-N
<b>Formula:</b>	C14H19BrO2
<b>SMILES:</b>	CCc1cccc(OC(=O)CCCCBr)c1
<b>Mol. weight [g/mol]:</b>	299.20

## Physical Properties

Property code	Value	Unit	Source
gf	-49.82	kJ/mol	Joback Method
hf	-325.70	kJ/mol	Joback Method
hfus	33.74	kJ/mol	Joback Method
hvap	65.29	kJ/mol	Joback Method
log10ws	-4.69		Crippen Method
logp	4.110		Crippen Method
mcvol	209.300	ml/mol	McGowan Method
pc	2202.08	kPa	Joback Method
rinpol	2093.00		NIST Webbook
rinpol	2093.00		NIST Webbook
tb	693.83	K	Joback Method
tc	905.84	K	Joback Method
tf	418.44	K	Joback Method
vc	0.797	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	536.57	J/molxK	693.83	Joback Method
cpg	601.62	J/molxK	870.50	Joback Method
cpg	590.31	J/molxK	835.17	Joback Method
cpg	578.19	J/molxK	799.83	Joback Method
cpg	565.21	J/molxK	764.50	Joback Method
cpg	551.35	J/molxK	729.16	Joback Method
cpg	612.15	J/molxK	905.84	Joback Method
dvisc	0.0001352	Paxs	693.83	Joback Method

dvisc	0.0001703	Paxs	647.93	Joback Method
dvisc	0.0002223	Paxs	602.03	Joback Method
dvisc	0.0003032	Paxs	556.13	Joback Method
dvisc	0.0004373	Paxs	510.24	Joback Method
dvisc	0.0006780	Paxs	464.34	Joback Method
dvisc	0.0011575	Paxs	418.44	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354718&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354718&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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