

1-Aminocyclopentanecarboxylic acid, 3-chloropropoxycarbonyl-, heptyl ester

Inchi: InChI=1S/C17H30ClNO4/c1-2-3-4-5-8-13-22-15(20)17(10-6-7-11-17)19-16(21)23-14-9-1
InchiKey: KPJIYUYOZUIECT-UHFFFAOYSA-N
Formula: C17H30ClNO4
SMILES: CCCCCCOC(=O)C1(NC(=O)OCCCCI)CCCC1
Mol. weight [g/mol]: 347.88

Physical Properties

Property code	Value	Unit	Source
gf	-267.06	kJ/mol	Joback Method
hf	-770.36	kJ/mol	Joback Method
hfus	42.29	kJ/mol	Joback Method
hvap	81.67	kJ/mol	Joback Method
log10ws	-4.99		Crippen Method
logp	4.168		Crippen Method
mvol	276.630	ml/mol	McGowan Method
pc	1529.46	kPa	Joback Method
rinpol	2362.00		NIST Webbook
rinpol	2362.00		NIST Webbook
tb	844.06	K	Joback Method
tc	1047.37	K	Joback Method
tf	543.05	K	Joback Method
vc	1.058	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	870.73	J/molxK	844.06	Joback Method
cpg	888.28	J/molxK	877.94	Joback Method
cpg	905.23	J/molxK	911.83	Joback Method
cpg	921.67	J/molxK	945.71	Joback Method
cpg	937.70	J/molxK	979.60	Joback Method
cpg	953.42	J/molxK	1013.48	Joback Method
cpg	968.93	J/molxK	1047.37	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U392625&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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