

Succinic acid, 1,1,1-trifluoroprop-2-yl heptadecyl ester

Inchi:	InChI=1S/C24H43F3O4/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-20-30-22(28)18-19-2
InchiKey:	HRFQQHIRFGGXGP-UHFFFAOYSA-N
Formula:	C24H43F3O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)CCC(=O)OC(C)C(F)(F)F
Mol. weight [g/mol]:	452.59

Physical Properties

Property code	Value	Unit	Source
gf	-900.67	kJ/mol	Joback Method
hf	-1630.65	kJ/mol	Joback Method
hfus	61.79	kJ/mol	Joback Method
hvap	83.19	kJ/mol	Joback Method
log10ws	-8.37		Crippen Method
logp	7.675		Crippen Method
mvol	369.210	ml/mol	McGowan Method
pc	798.89	kPa	Joback Method
rinpol	2562.00		NIST Webbook
rinpol	2562.00		NIST Webbook
tb	895.24	K	Joback Method
tc	1099.65	K	Joback Method
tf	493.75	K	Joback Method
vc	1.464	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1239.26	J/mol×K	895.24	Joback Method
cpg	1259.04	J/mol×K	929.31	Joback Method
cpg	1277.45	J/mol×K	963.38	Joback Method
cpg	1294.54	J/mol×K	997.45	Joback Method
cpg	1310.37	J/mol×K	1031.51	Joback Method
cpg	1324.99	J/mol×K	1065.58	Joback Method
cpg	1338.46	J/mol×K	1099.65	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U390887&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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