

# Disulfide, bis(4-methoxyphenyl)

<b>Other names:</b>	Anisole, 4,4'-dithiodi-p-Methoxyphenyl disulfide Bis(p-methoxyphenyl)disulfide Bis(4-methoxyphenyl) disulfide Disulfide, bis(p-methoxyphenyl) 4-Methoxyphenyl disulfide Di(4-methoxyphenyl)disulfide
<b>Inchi:</b>	InChI=1S/C14H14O2S2/c1-15-11-3-7-13(8-4-11)17-18-14-9-5-12(16-2)6-10-14/h3-10H,1
<b>InchiKey:</b>	PZQGLCGLPMWYBT-UHFFFAOYSA-N
<b>Formula:</b>	C14H14O2S2
<b>SMILES:</b>	COc1ccc(SSc2ccc(OC)cc2)cc1
<b>Mol. weight [g/mol]:</b>	278.39
<b>CAS:</b>	5335-87-5

## Physical Properties

Property code	Value	Unit	Source
gf	128.80	kJ/mol	Joback Method
hf	-62.87	kJ/mol	Joback Method
hfus	29.96	kJ/mol	Joback Method
hvap	71.09	kJ/mol	Joback Method
ie	7.60	eV	NIST Webbook
log10ws	-5.00		Crippen Method
logp	4.503		Crippen Method
mcvol	205.040	ml/mol	McGowan Method
pc	2684.64	kPa	Joback Method
tb	765.44	K	Joback Method
tc	1032.04	K	Joback Method
tf	438.68	K	Joback Method
vc	0.748	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	528.13	J/mol×K	765.44	Joback Method

cpg	542.85	J/mol×K	809.87	Joback Method
cpg	556.05	J/mol×K	854.31	Joback Method
cpg	567.73	J/mol×K	898.74	Joback Method
cpg	577.89	J/mol×K	943.17	Joback Method
cpg	586.52	J/mol×K	987.61	Joback Method
cpg	593.61	J/mol×K	1032.04	Joback Method

## Sources

NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5335875&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5335875&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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