

# Alloxan

<b>Other names:</b>	2,4,5,6(1H,3H)-Pyrimidinetetrone Alloxan 7169 Alloxane Barbituric acid, 5-oxo- Mesoxalylcarbamide Mesoxalylurea Urea, mesoxalyl- 2,4,5,6-Pyrimidintetrone 2,4,5,6-Tetraoxohexahydropyrimidine 2,4,5,6-Pyrimidinetetrone Pyrimidinetetrone NSC 7169
<b>Inchi:</b>	InChI=1S/C4H2N2O4/c7-1-2(8)5-4(10)6-3(1)9/h(H2,5,6,8,9,10)
<b>InchiKey:</b>	HIMXGTXNXJYFGB-UHFFFAOYSA-N
<b>Formula:</b>	C4H2N2O4
<b>SMILES:</b>	O=C1NC(=O)C(=O)C(=O)N1
<b>Mol. weight [g/mol]:</b>	142.07
<b>CAS:</b>	50-71-5

## Physical Properties

Property code	Value	Unit	Source
gf	-299.98	kJ/mol	Joback Method
hf	-526.41	kJ/mol	Joback Method
hfus	14.10	kJ/mol	Joback Method
hvap	55.74	kJ/mol	Joback Method
log10ws	0.65		Crippen Method
logp	-2.079		Crippen Method
mvol	82.600	ml/mol	McGowan Method
pc	7255.44	kPa	Joback Method
ss	153.00	J/molxK	NIST Webbook
tb	683.52	K	Joback Method
tc	977.54	K	Joback Method
tf	629.40	K	Joback Method
vc	0.295	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	209.39	J/mol×K	683.52	Joback Method
cpg	221.70	J/mol×K	732.52	Joback Method
cpg	233.01	J/mol×K	781.53	Joback Method
cpg	243.00	J/mol×K	830.53	Joback Method
cpg	251.37	J/mol×K	879.53	Joback Method
cpg	257.81	J/mol×K	928.54	Joback Method
cpg	261.99	J/mol×K	977.54	Joback Method
cps	153.01	J/mol×K	297.20	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C50715&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C50715&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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