

# Succinic acid, 3-methylbut-2-yl 2,4,6-trichlorophenyl ester

<b>Inchi:</b>	InChI=1S/C15H17Cl3O4/c1-8(2)9(3)21-13(19)4-5-14(20)22-15-11(17)6-10(16)7-12(15)18
<b>InchiKey:</b>	PQVFDIAUYQHIOI-UHFFFAOYSA-N
<b>Formula:</b>	C15H17Cl3O4
<b>SMILES:</b>	CC(C)C(C)OC(=O)CCC(=O)Oc1c(Cl)cc(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	367.65

## Physical Properties

Property code	Value	Unit	Source
gf	-349.57	kJ/mol	Joback Method
hf	-698.19	kJ/mol	Joback Method
hfus	38.60	kJ/mol	Joback Method
hvap	83.94	kJ/mol	Joback Method
log10ws	-5.50		Crippen Method
logp	4.920		Crippen Method
mcvol	250.050	ml/mol	McGowan Method
pc	1772.85	kPa	Joback Method
rinpol	2241.00		NIST Webbook
rinpol	2241.00		NIST Webbook
tb	848.21	K	Joback Method
tc	1069.88	K	Joback Method
tf	526.87	K	Joback Method
vc	0.951	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	665.82	J/molxK	848.21	Joback Method
cpg	677.35	J/molxK	885.15	Joback Method
cpg	687.83	J/molxK	922.10	Joback Method
cpg	697.25	J/molxK	959.04	Joback Method
cpg	705.62	J/molxK	995.99	Joback Method
cpg	712.96	J/molxK	1032.93	Joback Method
cpg	719.26	J/molxK	1069.88	Joback Method
dvisc	0.0005069	Paxs	526.87	Joback Method

dvisc	0.0003035	Paxs	580.43	Joback Method
dvisc	0.0001982	Paxs	633.98	Joback Method
dvisc	0.0001383	Paxs	687.54	Joback Method
dvisc	0.0001016	Paxs	741.10	Joback Method
dvisc	0.0000779	Paxs	794.65	Joback Method
dvisc	0.0000617	Paxs	848.21	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390271&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390271&amp;Units=SI</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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