

# Propionic acid, 2,3-dichloro-2-chloromethyl-, methyl ester

Inchi:	InChI=1S/C5H7Cl3O2/c1-10-4(9)5(8,2-6)3-7/h2-3H2,1H3
InchiKey:	KLXRARKCCAQPMU-UHFFFAOYSA-N
Formula:	C5H7Cl3O2
SMILES:	COC(=O)C(Cl)(CCl)CCl
Mol. weight [g/mol]:	205.47
CAS:	116594-64-0

## Physical Properties

Property code	Value	Unit	Source
gf	-275.65	kJ/mol	Joback Method
hf	-447.30	kJ/mol	Joback Method
hfus	16.67	kJ/mol	Joback Method
hvap	47.74	kJ/mol	Joback Method
log10ws	-1.35		Crippen Method
logp	1.615		Crippen Method
mcvol	125.470	ml/mol	McGowan Method
pc	3287.81	kPa	Joback Method
tb	499.15	K	Joback Method
tc	710.20	K	Joback Method
tf	310.45	K	Joback Method
vc	0.475	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	234.65	J/molxK	499.15	Joback Method
cpg	243.00	J/molxK	534.32	Joback Method
cpg	250.81	J/molxK	569.50	Joback Method
cpg	258.10	J/molxK	604.67	Joback Method
cpg	264.90	J/molxK	639.85	Joback Method
cpg	271.22	J/molxK	675.02	Joback Method
cpg	277.09	J/molxK	710.20	Joback Method
dvisc	0.0032418	Paxs	310.45	Joback Method
dvisc	0.0018441	Paxs	341.90	Joback Method

dvisc	0.0011536	Paxs	373.35	Joback Method
dvisc	0.0007762	Paxs	404.80	Joback Method
dvisc	0.0005530	Paxs	436.25	Joback Method
dvisc	0.0004123	Paxs	467.70	Joback Method
dvisc	0.0003190	Paxs	499.15	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C116594640&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C116594640&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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