

# 2-Naphthalenecarboxamide, N-(5-chloro-2,4-dimethoxyphenyl)-3-hydroxy-

Other names:

Acco Naf-Sol AS-ITR  
Acco Naphthol AS-ITR  
Acna Naphthol SS  
Amanil Naphthol AS-ITR  
Azoground ITR  
Azotol O  
C.I. Azoic Coupling Component 12  
C.I. 37550  
Daito Grounder ITR  
Hiltonaphthol AS-ITR  
Naphtanilide ITR  
Naphtazol STR  
Naphthoide ITR  
Naphthol AS-ITR  
Naphtoelan JTR  
Naphtol AS-ITR  
Naphtol AS-ITRL  
Sanatol ITR  
Solunaptol ITR  
Ultrazol IX-ITR  
2-Naphthanilide, 5'-chloro-3-hydroxy-2',4'-dimethoxy-  
5'-Chloro-3-hydroxy-2',4'-dimethoxy-2-naphthanilide  
NSC 50687

Inchi: InChI=1S/C19H16CINO4/c1-24-17-10-18(25-2)15(9-14(17)20)21-19(23)13-7-11-5-3-4-6-

InchiKey: XDWATWCCUTYUDE-UHFFFAOYSA-N

Formula: C19H16CINO4

SMILES: COc1cc(OC)c(NC(=O)c2cc3cccc3cc2O)cc1Cl

Mol. weight [g/mol]: 357.79

CAS: 92-72-8

## Physical Properties

Property code	Value	Unit	Source
gf	-14.03	kJ/mol	Joback Method
hf	-333.84	kJ/mol	Joback Method
hfus	47.56	kJ/mol	Joback Method
hvap	102.13	kJ/mol	Joback Method

log10ws	-5.72		Crippen Method
logp	4.468		Crippen Method
mcvol	252.990	ml/mol	McGowan Method
pc	2400.57	kPa	Joback Method
tb	993.31	K	Joback Method
tc	1245.42	K	Joback Method
tf	728.20	K	Joback Method
vc	0.897	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	751.24	J/mol×K	993.31	Joback Method
cpg	763.04	J/mol×K	1035.33	Joback Method
cpg	774.31	J/mol×K	1077.35	Joback Method
cpg	785.17	J/mol×K	1119.37	Joback Method
cpg	795.74	J/mol×K	1161.39	Joback Method
cpg	806.15	J/mol×K	1203.40	Joback Method
cpg	816.53	J/mol×K	1245.42	Joback Method

## Sources

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C92728&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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