

# Succinic acid, 2,3-dichlorophenyl neopentyl ester

<b>Inchi:</b>	InChI=1S/C15H18Cl2O4/c1-15(2,3)9-20-12(18)7-8-13(19)21-11-6-4-5-10(16)14(11)17/h4
<b>InchiKey:</b>	KIAUQVRRYUIUEZ-UHFFFAOYSA-N
<b>Formula:</b>	C15H18Cl2O4
<b>SMILES:</b>	CC(C)(C)COC(=O)CCC(=O)Oc1cccc(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	333.21

## Physical Properties

Property code	Value	Unit	Source
gf	-320.29	kJ/mol	Joback Method
hf	-669.17	kJ/mol	Joback Method
hfus	34.42	kJ/mol	Joback Method
hvap	78.37	kJ/mol	Joback Method
log10ws	-4.71		Crippen Method
logp	4.268		Crippen Method
mcvol	237.810	ml/mol	McGowan Method
pc	1857.91	kPa	Joback Method
rinpol	2207.00		NIST Webbook
rinpol	2207.00		NIST Webbook
tb	803.45	K	Joback Method
tc	1023.67	K	Joback Method
tf	516.85	K	Joback Method
vc	0.902	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	645.24	J/molxK	803.45	Joback Method
cpg	698.73	J/molxK	986.97	Joback Method
cpg	689.94	J/molxK	950.26	Joback Method
cpg	680.23	J/molxK	913.56	Joback Method
cpg	669.56	J/molxK	876.86	Joback Method
cpg	657.91	J/molxK	840.15	Joback Method
cpg	706.62	J/molxK	1023.67	Joback Method
dvisc	0.0000671	Paxs	803.45	Joback Method

dvisc	0.0000848	Paxs	755.68	Joback Method
dvisc	0.0001106	Paxs	707.92	Joback Method
dvisc	0.0001498	Paxs	660.15	Joback Method
dvisc	0.0002129	Paxs	612.38	Joback Method
dvisc	0.0003211	Paxs	564.62	Joback Method
dvisc	0.0005224	Paxs	516.85	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389589&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389589&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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