

# 1,2-Benzenediol, 4,5-dichloro-

<b>Other names:</b>	Pyrocatechol, 4,5-dichloro- 4,5-Dichlorocatechol 4,5-Dichloropyrocatechol 4,5-Dichloro-benzene-1,2-diol 4,5-Dichloro-1,2-benzenediol
<b>Inchi:</b>	InChI=1S/C6H4Cl2O2/c7-3-1-5(9)6(10)2-4(3)8/h1-2,9-10H
<b>InchiKey:</b>	ACCHWUWBKYGKNM-UHFFFAOYSA-N
<b>Formula:</b>	C6H4Cl2O2
<b>SMILES:</b>	Oc1cc(Cl)c(Cl)cc1O
<b>Mol. weight [g/mol]:</b>	179.00
<b>CAS:</b>	3428-24-8

## Physical Properties

Property code	Value	Unit	Source
gf	-230.68	kJ/mol	Joback Method
hf	-328.21	kJ/mol	Joback Method
hfus	24.91	kJ/mol	Joback Method
hvap	66.69	kJ/mol	Joback Method
log10ws	-1.94		Crippen Method
logp	2.405		Crippen Method
mcvol	107.860	ml/mol	McGowan Method
pc	6359.24	kPa	Joback Method
tb	604.44	K	Joback Method
tc	861.92	K	Joback Method
tf	479.60	K	Joback Method
vc	0.293	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	250.80	J/molxK	861.92	Joback Method
cpg	220.56	J/molxK	604.44	Joback Method
cpg	226.43	J/molxK	647.35	Joback Method
cpg	231.76	J/molxK	690.27	Joback Method

cpg	236.70	J/molxK	733.18	Joback Method
cpg	241.40	J/molxK	776.09	Joback Method
cpg	246.05	J/molxK	819.00	Joback Method
dvisc	0.0000081	Paxs	604.44	Joback Method
dvisc	0.0001179	Paxs	479.60	Joback Method
dvisc	0.0000687	Paxs	500.41	Joback Method
dvisc	0.0000418	Paxs	521.21	Joback Method
dvisc	0.0000264	Paxs	542.02	Joback Method
dvisc	0.0000173	Paxs	562.83	Joback Method
dvisc	0.0000117	Paxs	583.63	Joback Method
hvapt	70.50	kJ/mol	308.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3428248&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3428248&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/121-203-1/1-2-Benzenediol-4-5-dichloro.pdf>

Generated by Cheméo on 2023-03-26 03:42:43.128312064 +0000 UTC m=+847531.023436088.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.