

# 1,2-Cyclohexanedicarboxylic acid, 2,5-dichlorophenyl isobutyl ester

Inchi:	InChI=1S/C18H22Cl2O4/c1-11(2)10-23-17(21)13-5-3-4-6-14(13)18(22)24-16-9-12(19)7-8
InchiKey:	PRZLLEOYJKKUJN-UHFFFAOYSA-N
Formula:	C18H22Cl2O4
SMILES:	CC(C)COC(=O)C1CCCCC1C(=O)Oc1cc(Cl)ccc1Cl
Mol. weight [g/mol]:	373.27

## Physical Properties

Property code	Value	Unit	Source
gf	-283.57	kJ/mol	Joback Method
hf	-693.64	kJ/mol	Joback Method
hfus	38.99	kJ/mol	Joback Method
hvap	86.08	kJ/mol	Joback Method
log10ws	-5.37		Crippen Method
logp	4.904		Crippen Method
mcvol	269.220	ml/mol	McGowan Method
pc	1670.06	kPa	Joback Method
rinpol	2485.00		NIST Webbook
rinpol	2485.00		NIST Webbook
tb	889.76	K	Joback Method
tc	1122.99	K	Joback Method
tf	536.38	K	Joback Method
vc	1.008	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	810.82	J/mol×K	889.76	Joback Method
cpg	866.57	J/mol×K	1084.12	Joback Method
cpg	858.52	J/mol×K	1045.25	Joback Method
cpg	848.93	J/mol×K	1006.37	Joback Method
cpg	837.80	J/mol×K	967.50	Joback Method
cpg	825.10	J/mol×K	928.63	Joback Method
cpg	873.12	J/mol×K	1122.99	Joback Method
dvisc	0.0000679	Paxs	889.76	Joback Method

dvisc	0.0000858	Paxs	830.86	Joback Method
dvisc	0.0001123	Paxs	771.97	Joback Method
dvisc	0.0001536	Paxs	713.07	Joback Method
dvisc	0.0002224	Paxs	654.17	Joback Method
dvisc	0.0003464	Paxs	595.28	Joback Method
dvisc	0.0005947	Paxs	536.38	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339797&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339797&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rlnpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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