

Benzoic acid, 4-(dipropylamino)-, methyl ester

Inchi:

InchiKey:

Formula:

SMILES:

Mol. weight [g/mol]:

InChI=1S/C14H21NO2/c1-4-10-15(11-5-2)13-8-6-12(7-9-13)14(16)17-3/h6-9H,4-5,10-11H

HEZNCGYOOCSLH-UHFFFAOYSA-N

C14H21NO2

CCCN(CCC)c1ccc(C(=O)OC)cc1

235.32

Physical Properties

Property code	Value	Unit	Source
gf	46.64	kJ/mol	Joback Method
hf	-284.50	kJ/mol	Joback Method
hfus	31.48	kJ/mol	Joback Method
hvap	60.90	kJ/mol	Joback Method
log10ws	-3.30		Crippen Method
logp	3.100		Crippen Method
mcvol	201.780	ml/mol	McGowan Method
pc	2062.36	kPa	Joback Method
rinpol	1959.00		NIST Webbook
rinpol	1959.00		NIST Webbook
tb	640.11	K	Joback Method
tc	838.70	K	Joback Method
tf	391.11	K	Joback Method
vc	0.753	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	532.25	J/mol×K	640.11	Joback Method
cpg	548.68	J/mol×K	673.21	Joback Method
cpg	564.18	J/mol×K	706.31	Joback Method
cpg	578.78	J/mol×K	739.41	Joback Method
cpg	592.51	J/mol×K	772.51	Joback Method
cpg	605.40	J/mol×K	805.60	Joback Method
cpg	617.48	J/mol×K	838.70	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U374482&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinsol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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