

# L-Valine, N-(2,6-difluorobenzoyl)-, hexyl ester

<b>Inchi:</b>	InChI=1S/C18H25F2NO3/c1-4-5-6-7-11-24-18(23)16(12(2)3)21-17(22)15-13(19)9-8-10-1
<b>InchiKey:</b>	PURSPZPABDNROW-UHFFFAOYSA-N
<b>Formula:</b>	C18H25F2NO3
<b>SMILES:</b>	CCCCCOC(=O)C(NC(=O)c1c(F)cccc1F)C(C)C
<b>Mol. weight [g/mol]:</b>	341.39

## Physical Properties

Property code	Value	Unit	Source
gf	-474.12	kJ/mol	Joback Method
hf	-907.95	kJ/mol	Joback Method
hfus	44.24	kJ/mol	Joback Method
hvap	79.19	kJ/mol	Joback Method
log10ws	-5.40		Crippen Method
logp	3.843		Crippen Method
mvol	263.250	ml/mol	McGowan Method
pc	1482.71	kPa	Joback Method
rinpol	2270.00		NIST Webbook
rinpol	2270.00		NIST Webbook
tb	825.87	K	Joback Method
tc	1023.86	K	Joback Method
tf	490.01	K	Joback Method
vc	1.024	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	805.88	J/mol×K	825.87	Joback Method
cpg	820.52	J/mol×K	858.87	Joback Method
cpg	834.14	J/mol×K	891.87	Joback Method
cpg	846.78	J/mol×K	924.86	Joback Method
cpg	858.45	J/mol×K	957.86	Joback Method
cpg	869.19	J/mol×K	990.86	Joback Method
cpg	879.03	J/mol×K	1023.86	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U346619&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U346619&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvpap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinppl:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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