

# Ethylenediamine-N,N'-disuccinic acid

<b>Inchi:</b>	InChI=1S/C10H16N2O8/c13-7(14)3-5(9(17)18)11-1-2-12-6(10(19)20)4-8(15)16/h5-6,11-
<b>InchiKey:</b>	VKZRWSNIWNFCIQ-UHFFFAOYSA-N
<b>Formula:</b>	C10H16N2O8
<b>SMILES:</b>	O=C(O)CC(NCCNC(CC(=O)O)C(=O)O)C(=O)O
<b>Mol. weight [g/mol]:</b>	292.24
<b>CAS:</b>	20846-91-7

## Physical Properties

Property code	Value	Unit	Source
chs	-4271.60 ± 3.90	kJ/mol	NIST Webbook
chs	-4252.10 ± 4.20	kJ/mol	NIST Webbook
gf	-855.74	kJ/mol	Joback Method
hf	-1212.59	kJ/mol	Joback Method
hfs	-1969.40 ± 4.20	kJ/mol	NIST Webbook
hfs	-1950.20 ± 3.90	kJ/mol	NIST Webbook
hfus	47.56	kJ/mol	Joback Method
hvap	143.65	kJ/mol	Joback Method
log10ws	0.99		Crippen Method
logp	-1.979		Crippen Method
mcvol	201.480	ml/mol	McGowan Method
pc	4351.13	kPa	Joback Method
tb	1111.86	K	Joback Method
tc	1391.50	K	Joback Method
tf	720.78	K	Joback Method
vc	0.753	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	659.18	J/mol×K	1111.86	Joback Method
cpg	665.03	J/mol×K	1158.47	Joback Method
cpg	669.93	J/mol×K	1205.07	Joback Method
cpg	673.98	J/mol×K	1251.68	Joback Method
cpg	677.25	J/mol×K	1298.29	Joback Method

cpg	679.82	J/mol×K	1344.89	Joback Method
cpg	681.77	J/mol×K	1391.50	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20846917&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20846917&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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