

DL-Alanine, N-methyl-N-(3-chloropropoxycarbonyl)-, pentadecyl ester

InChI: InChI=1S/C23H44ClNO4/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-19-28-22(26)21(2)25(3)2
InChIKey: NRLOBYKKNONJHMO-UHFFFAOYSA-N

Formula: C23H44ClNO4

SMILES: CCCCCCCCCCCCCCOC(=O)C(C)N(C)C(=O)OCCCCI

Mol. weight [g/mol]: 434.05

Physical Properties

Property code	Value	Unit	Source
gf	-228.65	kJ/mol	Joback Method
hf	-961.14	kJ/mol	Joback Method
hfus	64.60	kJ/mol	Joback Method
hvap	91.14	kJ/mol	Joback Method
log10ws	-6.99		Crippen Method
logp	6.707		Crippen Method
mvol	372.030	ml/mol	McGowan Method
pc	883.14	kPa	Joback Method
rinpol	2868.00		NIST Webbook
rinpol	2868.00		NIST Webbook
tb	927.65	K	Joback Method
tc	1137.36	K	Joback Method
tf	540.68	K	Joback Method
vc	1.433	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1230.29	J/molxK	927.65	Joback Method
cpg	1249.17	J/molxK	962.60	Joback Method
cpg	1266.63	J/molxK	997.55	Joback Method
cpg	1282.71	J/molxK	1032.51	Joback Method
cpg	1297.46	J/molxK	1067.46	Joback Method
cpg	1310.93	J/molxK	1102.41	Joback Method
cpg	1323.16	J/molxK	1137.36	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U392787&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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