

# Benzenamine, N,N'-1,2-ethanediylidenebis[4-methoxy-

Other names:

Benzenamine, N,N'-1,2-ethanediylidenebis\*4-methoxy-  
N,N'-Ethanediylidenebis[4-methoxybenzenamine]

Inchi:

InChI=1S/C16H16N2O2/c1-19-15-7-3-13(4-8-15)17-11-12-18-14-5-9-16(20-2)10-6-14/h3

InchiKey:

IRMGPOUTEGABKC-UHFFFAOYSA-N

Formula:

C16H16N2O2

SMILES:

COc1ccc(N=CC=Nc2ccc(OC)cc2)cc1

Mol. weight [g/mol]:

268.31

CAS:

24764-91-8

## Physical Properties

Property code	Value	Unit	Source
hf	-23.45	kJ/mol	Joback Method
hvap	68.53	kJ/mol	Joback Method
ie	7.10	eV	NIST Webbook
ie	7.75	eV	NIST Webbook
log10ws	-3.66		Crippen Method
logp	3.809		Crippen Method
mvol	211.880	ml/mol	McGowan Method
pc	1786.37	kPa	Joback Method
tb	827.00	K	Joback Method
tc	1076.93	K	Joback Method

## Sources

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C24764918&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>ws:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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