

# Succinic acid, 2,2-dichloroethyl 2-methoxy-5-methylphenyl ester

Inchi:	InChI=1S/C14H16Cl2O5/c1-9-3-4-10(19-2)11(7-9)21-14(18)6-5-13(17)20-8-12(15)16/h3-
InchiKey:	LDQOWQUHQJICEL-UHFFFAOYSA-N
Formula:	C14H16Cl2O5
SMILES:	COc1ccc(C)cc1OC(=O)CCC(=O)OCC(Cl)Cl
Mol. weight [g/mol]:	335.18

## Physical Properties

Property code	Value	Unit	Source
gf	-438.99	kJ/mol	Joback Method
hf	-777.28	kJ/mol	Joback Method
hfus	36.91	kJ/mol	Joback Method
hvap	79.46	kJ/mol	Joback Method
log10ws	-3.83		Crippen Method
logp	3.036		Crippen Method
mvol	229.590	ml/mol	McGowan Method
pc	1971.80	kPa	Joback Method
rinpol	2319.00		NIST Webbook
rinpol	2319.00		NIST Webbook
tb	805.78	K	Joback Method
tc	1021.07	K	Joback Method
tf	510.39	K	Joback Method
vc	0.870	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	615.26	J/molxK	805.78	Joback Method
cpg	627.20	J/molxK	841.66	Joback Method
cpg	638.12	J/molxK	877.54	Joback Method
cpg	648.01	J/molxK	913.43	Joback Method
cpg	656.87	J/molxK	949.31	Joback Method
cpg	664.69	J/molxK	985.19	Joback Method
cpg	671.46	J/molxK	1021.07	Joback Method
dvisc	0.0004965	Paxs	510.39	Joback Method

dvisc	0.0003065	Paxs	559.62	Joback Method
dvisc	0.0002046	Paxs	608.85	Joback Method
dvisc	0.0001450	Paxs	658.09	Joback Method
dvisc	0.0001079	Paxs	707.32	Joback Method
dvisc	0.0000834	Paxs	756.55	Joback Method
dvisc	0.0000665	Paxs	805.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390957&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390957&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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