

Benzamide, 2,5-difluoro-N-hexadecyl-

Inchi:	InChI=1S/C23H37F2NO/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-18-26-23(27)21-19-20(24)
InchiKey:	AYZZXPKDUBZPCX-UHFFFAOYSA-N
Formula:	C23H37F2NO
SMILES:	CCCCCCCCCCCCCCCCNC(=O)c1cc(F)ccc1F
Mol. weight [g/mol]:	381.54

Physical Properties

Property code	Value	Unit	Source
gf	-193.22	kJ/mol	Joback Method
hf	-755.79	kJ/mol	Joback Method
hfus	61.45	kJ/mol	Joback Method
hvap	81.94	kJ/mol	Joback Method
log10ws	-8.76		Crippen Method
logp	7.176		Crippen Method
mvol	326.260	ml/mol	McGowan Method
pc	1014.89	kPa	Joback Method
rmpol	2799.00		NIST Webbook
rmpol	2799.00		NIST Webbook
tb	864.86	K	Joback Method
tc	1059.83	K	Joback Method
tf	504.20	K	Joback Method
vc	1.292	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1053.23	J/molxK	864.86	Joback Method
cpg	1071.26	J/molxK	897.35	Joback Method
cpg	1088.21	J/molxK	929.85	Joback Method
cpg	1104.13	J/molxK	962.34	Joback Method
cpg	1119.08	J/molxK	994.84	Joback Method
cpg	1133.10	J/molxK	1027.33	Joback Method
cpg	1146.24	J/molxK	1059.83	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U407596&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvpap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinppl:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/121-526-3/Benzamide-2-5-difluoro-N-hexadecyl.pdf>

Generated by Cheméo on 2024-05-01 10:07:05.090256975 +0000 UTC m=+16847274.010834287.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.