

# Fumaric acid, 2-phenethyl 2,4,6-trichlorophenyl ester

**Inchi:** InChI=1S/C18H13Cl3O4/c19-13-10-14(20)18(15(21)11-13)25-17(23)7-6-16(22)24-9-8-12  
**InchiKey:** KEKSIWZMPRYBRX-VOTSOKGWSA-N  
**Formula:** C18H13Cl3O4  
**SMILES:** O=C(C=CC(=O)Oc1c(Cl)cc(Cl)cc1Cl)OCCc1ccccc1  
**Mol. weight [g/mol]:** 399.65

## Physical Properties

Property code	Value	Unit	Source
gf	-126.80	kJ/mol	Joback Method
hf	-395.80	kJ/mol	Joback Method
hfus	47.66	kJ/mol	Joback Method
hvap	93.62	kJ/mol	Joback Method
log10ws	-5.85		Crippen Method
logp	4.894		Crippen Method
mcvol	264.260	ml/mol	McGowan Method
pc	1896.95	kPa	Joback Method
rinpol	2788.00		NIST Webbook
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tb	948.57	K	Joback Method
tc	1193.72	K	Joback Method
tf	612.02	K	Joback Method
vc	1.002	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	699.44	J/molxK	948.57	Joback Method
cpg	708.82	J/molxK	989.43	Joback Method
cpg	717.10	J/molxK	1030.29	Joback Method
cpg	724.34	J/molxK	1071.14	Joback Method
cpg	730.58	J/molxK	1112.00	Joback Method
cpg	735.88	J/molxK	1152.86	Joback Method
cpg	740.30	J/molxK	1193.72	Joback Method
dvisc	0.0002686	Paxs	612.02	Joback Method

dvisc	0.0001744	Paxs	668.11	Joback Method
dvisc	0.0001211	Paxs	724.20	Joback Method
dvisc	0.0000887	Paxs	780.29	Joback Method
dvisc	0.0000677	Paxs	836.39	Joback Method
dvisc	0.0000534	Paxs	892.48	Joback Method
dvisc	0.0000434	Paxs	948.57	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405685&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405685&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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