

# Malonic acid, 2,2-dichloroethyl nonyl ester

<b>Inchi:</b>	InChI=1S/C14H24Cl2O4/c1-2-3-4-5-6-7-8-9-19-13(17)10-14(18)20-11-12(15)16/h12H,2-
<b>InchiKey:</b>	ZXSUKHKMKVFXIJ-UHFFFAOYSA-N
<b>Formula:</b>	C14H24Cl2O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)CC(=O)OCC(Cl)Cl
<b>Mol. weight [g/mol]:</b>	327.24

## Physical Properties

Property code	Value	Unit	Source
gf	-427.14	kJ/mol	Joback Method
hf	-858.65	kJ/mol	Joback Method
hfus	42.46	kJ/mol	Joback Method
hvap	73.45	kJ/mol	Joback Method
log10ws	-4.32		Crippen Method
logp	4.017		Crippen Method
mcvol	247.480	ml/mol	McGowan Method
pc	1554.90	kPa	Joback Method
rinpol	2059.00		NIST Webbook
rinpol	2059.00		NIST Webbook
tb	746.72	K	Joback Method
tc	934.24	K	Joback Method
tf	436.70	K	Joback Method
vc	0.960	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	681.87	J/molxK	746.72	Joback Method
cpg	744.64	J/molxK	902.99	Joback Method
cpg	733.69	J/molxK	871.74	Joback Method
cpg	721.94	J/molxK	840.48	Joback Method
cpg	709.39	J/molxK	809.23	Joback Method
cpg	696.04	J/molxK	777.97	Joback Method
cpg	754.82	J/molxK	934.24	Joback Method
dvisc	0.0000864	Paxs	746.72	Joback Method

dvisc	0.0001136	Paxs	695.05	Joback Method
dvisc	0.0001560	Paxs	643.38	Joback Method
dvisc	0.0002264	Paxs	591.71	Joback Method
dvisc	0.0003530	Paxs	540.04	Joback Method
dvisc	0.0006045	Paxs	488.37	Joback Method
dvisc	0.0011757	Paxs	436.70	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U349064&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U349064&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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