

# 4(1,1-Dimethylbutyl)phenol, pentafluorobenzoyl ester

**Inchi:** InChI=1S/C19H17F5O2/c1-4-9-19(2,3)10-5-7-11(8-6-10)26-18(25)12-13(20)15(22)17(24)  
**InchiKey:** OAZHPWDYKBFPII-UHFFFAOYSA-N  
**Formula:** C19H17F5O2  
**SMILES:** CCCC(C)(C)c1ccc(OC(=O)c2c(F)c(F)c(F)c(F)c2F)cc1  
**Mol. weight [g/mol]:** 372.33

## Physical Properties

Property code	Value	Unit	Source
gf	-928.99	kJ/mol	Joback Method
hf	-1265.35	kJ/mol	Joback Method
hfus	41.49	kJ/mol	Joback Method
hvap	70.19	kJ/mol	Joback Method
log10ws	-7.37		Crippen Method
logp	5.679		Crippen Method
mcvol	247.340	ml/mol	McGowan Method
pc	1445.73	kPa	Joback Method
rinpol	1984.40		NIST Webbook
rinpol	1992.60		NIST Webbook
rinpol	1987.80		NIST Webbook
rinpol	1984.40		NIST Webbook
tb	786.77	K	Joback Method
tc	986.29	K	Joback Method
tf	509.38	K	Joback Method
vc	0.987	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	716.98	J/molxK	786.77	Joback Method
cpg	730.52	J/molxK	820.02	Joback Method
cpg	743.14	J/molxK	853.28	Joback Method
cpg	754.86	J/molxK	886.53	Joback Method
cpg	765.73	J/molxK	919.78	Joback Method
cpg	775.76	J/molxK	953.04	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R433248&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R433248&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/121-743-2/4-1-1-Dimethylbutyl-phenol-pentafluorobenzoyl-ester.pdf>

Generated by Cheméo on 2024-04-27 17:48:45.54313094 +0000 UTC m=+16529374.463708251.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.