

# Succinic acid, 2,2-dichloroethyl 1-cyclopentylethyl ester

**Inchi:** InChI=1S/C13H20Cl2O4/c1-9(10-4-2-3-5-10)19-13(17)7-6-12(16)18-8-11(14)15/h9-11H,1-3H2  
**InchiKey:** QNNJUBWLHQJVOQ-UHFFFAOYSA-N  
**Formula:** C13H20Cl2O4  
**SMILES:** CC(OC(=O)CCC(=O)OCC(Cl)Cl)C1CCCC1  
**Mol. weight [g/mol]:** 311.20

## Physical Properties

Property code	Value	Unit	Source
gf	-401.45	kJ/mol	Joback Method
hf	-782.81	kJ/mol	Joback Method
hfus	30.28	kJ/mol	Joback Method
hvap	71.09	kJ/mol	Joback Method
log10ws	-3.67		Crippen Method
logp	3.235		Crippen Method
mvol	222.530	ml/mol	McGowan Method
pc	1971.80	kPa	Joback Method
rinpol	2045.00		NIST Webbook
rinpol	2045.00		NIST Webbook
tb	738.68	K	Joback Method
tc	949.03	K	Joback Method
tf	421.33	K	Joback Method
vc	0.839	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	617.65	J/molxK	738.68	Joback Method
cpg	683.02	J/molxK	913.97	Joback Method
cpg	671.99	J/molxK	878.91	Joback Method
cpg	659.96	J/molxK	843.86	Joback Method
cpg	646.90	J/molxK	808.80	Joback Method
cpg	632.81	J/molxK	773.74	Joback Method
cpg	693.07	J/molxK	949.03	Joback Method
dvisc	0.0001331	Paxs	738.68	Joback Method

dvisc	0.0001746	Paxs	685.79	Joback Method
dvisc	0.0002397	Paxs	632.90	Joback Method
dvisc	0.0003487	Paxs	580.00	Joback Method
dvisc	0.0005467	Paxs	527.11	Joback Method
dvisc	0.0009478	Paxs	474.22	Joback Method
dvisc	0.0018866	Paxs	421.33	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391411&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391411&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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